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1 ***Erato polymnioides*- a novel Hg hyperaccumulator plant in**
2 **ecuadorian rainforest acid soils with potential of microbe-**
3 **associated phytoremediation**

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HIGHLIGHTS:

- Hg accumulation capacity was studied in three plant species.
- Hg in acid soil studied is higher than the Ecuadorian threshold (0.1 mg kg^{-1}).
- Concentrations of Hg in roots and leaves were higher than in stems.
- All plants showed arbuscular mycorrhizal fungi colonization in their roots.
- *Erato polymnioides* showed high potential as an Hg hyperaccumulator.

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Abstract

Mercury (Hg) accumulation capacity was assessed in three plant species (*Axonopus compressus*, *Erato polymnioides*, and *Miconia zamorensis*) that grow on soils polluted by artisanal small-scale gold mines in the Ecuadorian rainforest. Individuals of three species were collected at two sampling zones: i) an intensive zone (IZ, 4.8 mg Hg kg⁻¹ of soil) where gold extraction continues to occur, and ii) a natural zone (NZ, 0.19 mg Hg kg⁻¹ of soil). In addition, the percentage of arbuscular mycorrhizal fungi (AMF) colonization was determined in plant roots and seven fungal morphotypes isolated from rhizospheric soil. Results suggest a facilitation role of native and pollution adapted AMF on Hg phytoaccumulation. E.g., *E. polymnioides* increased Hg accumulation when growing with greater AMF colonization. We concluded that *E. polymnioides* is a good candidate for the design of microbe-assisted strategies for Hg remediation at gold mining areas. The consortia between *E. polymnioides* and the AMF isolated in this study could be instrumental to get a deeper understanding of the AMF role in Hg phytoaccumulation.

Keywords: *Heavy metals; bioremediation; pollution; artisanal scale gold mining; arbuscular mycorrhizal fungi; southern Ecuador*

1. Introduction

Gold (Au) price has tripled in the last 10 years and artisanal small-scale gold mining (ASGM) has increased substantially in rural areas all over the world (García et al., 2015; Veiga et al., 2014). Around 16 million people are directly involved in this activity nowadays, producing 380-450 tons of gold annually. In many cases, the process used in ASGM for Au extraction is mercury (Hg) amalgamation, that consists in grinding the

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4 91 mineral (raw material) and mixing it with Hg (Seccatore et al., 2014). During
5 92 amalgamation, an Au-Hg alloy is produced and heated in open vessels to separate both
6 93 metals from undesired matter. Pure Au and volatilized Hg are then obtained.
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10 95 Hg is one of the most toxic heavy metals and a global pollutant that biomagnifies through
11 96 the food chain, posing a threat to human and animal health (Navarro et al., 2009; Pirrone et
12 97 al., 2009). ASGM using amalgamation is estimated to release into the environment 37% of
13 98 global air emissions of Hg and 1400 tons year⁻¹ of Hg directly to the soil and water sheds
14 99 (Gibb and O’Leary, 2014). It is also responsible for Hg emissions that lead to massive
15 100 aerial Hg contamination and pollution of terrestrial ecosystems through rainfall deposition
16 101 (727 tons year⁻¹) not only at ASGM sites but also at other areas (Seccatore et al., 2014;
17 102 United Nations Environment Programme, 2013). Anomalously high concentrations of Hg
18 103 (over 4 mg Hg kg⁻¹ soil) are reported by several studies wherever the Hg amalgamation
19 104 process is practiced (García et al., 2015; Terán-Mita et al., 2013). However, volatilization
20 105 of Hg from soil and water occurs and Hg returns to the atmosphere generating a cycle, with
21 106 volatilization from soil being more relevant than from the oceans (Bjerregaard and
22 107 Andersen, 2007). Plants contribute to volatilization of Hg by uptaking Hg from soil and
23 108 releasing it into the atmosphere through stomata in a process called phytovolatilization (Ali
24 109 et al., 2013).
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32 111 Communities living nearby and downstream ASGM are exposed to Hg vapor, and regularly
33 112 consume food heavily contaminated with methyl mercury (MeHg) (Gibb and O’Leary,
34 113 2014). Elemental Hg and MeHg are toxic to the nervous, digestive, and immune system.
35 114 They can cause mental retardation, seizures, vision and hearing loss, delayed development,
36 115 language disorders, and memory loss, being fatal in some cases (World Health
37 116 Organization, 2006). Reducing the impacts of ASGM on human health and the
38 117 environment has become a major concern for society and many governments worldwide.
39 118 Thus, reducing the use of Hg in ASGM was included in the Minamata Convention on Hg,
40 119 signed in 2013 in Minamanta, Japan (United Nations Environment Programme, 2013).
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46 121 Several techniques to encapsulate and stabilize Hg are being researched to reduce the
47 122 impact of Hg emissions at ASGM sites, such as bioremediation and, especially,
48 123 phytoremediation (Mani and Kumar, 2014). Hg phytoremediation with so called metal
49 124 hyperaccumulator plants has attracted interest as an inexpensive, low-impact and visually
50 125 benign technique compared to traditional physical approaches (Lorestani et al., 2012).
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55 127 Historically, the term "hyperaccumulator" has undergone significant changes since it was
56 128 defined by Brooks et al. (1977). The first definition was "plants with Ni concentrations
57 129 higher than 1000 mg kg⁻¹ dry weigh". Later, Baker and Brooks (1989) extended the concept
58 130 to more metals and defined hyperaccumulators as plant species which accumulate greater
59 131 than 100 mg kg⁻¹ dry weight Cd, or greater than 1000 mg kg⁻¹ dry weight Ni, Cu and Pb or
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132 greater than 10000 mg kg⁻¹ dry weight Zn and Mn in their shoots. However, other authors
133 suggest that criteria are unnecessarily conservative (van der Ent et al., 2013). Nowadays,
134 hyperaccumulators are defined as plants that achieve 100-fold higher shoot metal
135 concentration, compared to crop plants or common non-accumulator plants (Barceló and
136 Poschenrieder, 2003; Chamba et al., 2016; Redondo-Gómez, 2013). Thus, they have
137 tremendous potential for remediation of metals in the environment.

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139 Many plant species have been cataloged as Hg hyperaccumulators (Angle et al., 2001;
140 Lorestani et al., 2012), e.g. *Jatropha curcas* and *Piper marginatum* from tropical
141 rainforests and *Rumex induratus* and *Marrubium vulgare* from the Iberian Peninsula. *P.*
142 *marginatum* reached Hg tissue concentrations in the range of 0.53-6 mg kg⁻¹ (Marrugo-
143 Negrete et al., 2016) and those from the Iberian Peninsula showed 8.3-67.2 mg kg⁻¹ of Hg
144 in their roots (Moreno-Jiménez et al., 2006). Prasad and De Oliveira (2003) found
145 Asteraceae family members as one of the best hyperaccumulator plant species and a good
146 candidate for phytoremediation compared to several organism, including mycorrhizal and
147 non-mycorrhizal fungi, about 400 hyperaccumulating agricultural, vegetable crops and
148 ornamentals plants (Asteraceae, Brassicaceae, Caryophyllaceae, Cyperaceae,
149 Cunouniaceae, Fabaceae, Flacourtiaceae, Lamiaceae, Poaceae, Violaceae, and
150 Euphobiaceae).

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152 Finding native hyperaccumulator plants to the ASGM areas has gained attention during
153 recent years. Native metal hyperaccumulator plants are harmless to native biodiversity,
154 require less handling and acclimatization and are also, in most cases, highly tolerant to the
155 contaminants that need to be removed (Sarma, 2011). However, native hyperaccumulator
156 plants are difficult to cultivate sometimes, and may have low growth rates and biomass,
157 lowering the metal uptake (Goltapeh et al., 2013). In this sense, mycorrhizal fungi live
158 associated to most of the higher plants in different forms, with arbuscular mycorrhizal fungi
159 (AMF) associations with the roots of terrestrial plants being the most widespread. AMF
160 modify the bioavailability and mobility of toxic metals in the soil and facilitate metal
161 uptake by the plants as well as increase the host plants biomass production in polluted
162 ecosystems (Leung et al., 2013; Teixeira et al., 2014). AMF constitute a bridge for nutrient
163 and heavy metal transport from soils to plant roots, such as N, P, K, Ca, S, Zn, Co, Ni and
164 Cu through extensive hyphal network. They also bind metals in the hyphae outside the root,
165 preventing metals to move to the aerial parts of the plant and, in the case of Hg, preventing
166 volatilization too (Sarwar et al., 2017). Then, native AMF have arisen as promising
167 alternative and as an innovative tool to replace or supplement present treatment processes
168 in order to assist phytoremediation.

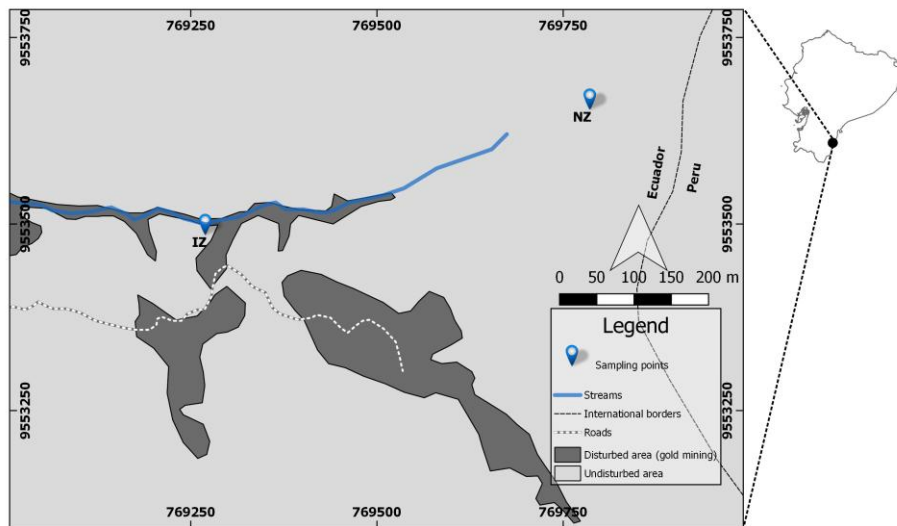
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170 This research has two main objectives. Firstly, to assess the potential of three native plants
171 as Hg hyperaccumulators in Chinapintza (Zamora-Chinchi province, southeast Ecuador),

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4 172 a site with a long ASGM tradition. Secondly, to evaluate AMF colonization in the plant
5 173 species and the spores of the most prominent AMF isolated.
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8 175 2. Materials and Methods

9 176 2.1. Study area

10 177 As shown in Figure 1, the study area is located in Chinapintza (Zamora-Chinchipe
11 178 Province, southeast Ecuador), beside the Condor mountain range, near to the Peruvian
12 179 border (at 1854 masl; 4°02'16.6"S 78°34'14.9"W). Zamora-Chinchipe Province is well
13 180 known by its ASGM activity: around 23% (282.998 ha) of the total surface is dedicated to
14 181 ASGM (Sacher and Báez, 2011). The study area was divided into two zones: (1) a natural
15 182 zone (NZ) free of influence of ASGM and (2) an intensive zone (IZ) highly deteriorated by
16 183 ASGM that continues to occur.



184
185 **Figure 1.** Map of the sampling points.

186 187 2.2. Plant and soil sampling and processing

188 Three abundant plants present in NZ and IZ were selected as potential Hg
189 hyperaccumulators: *Axonopus compressus* (SW.) P. Beauv (Poaceae), *Erato polymnioides*
190 DC. (Asteraceae), and *Miconia zamorensis* Gleason (Melastomataceae). Seven individuals
191 of each species and their rhizosphere soil were taken as samples in each of the two zones,
192 i.e. a total of 42 individuals and its surrounding soil.

193
194 According to the recommendations of Tack and Verloo (1996), sampling took place far
195 from active roads and the surface of the fresh plant material was checked to be free of dust.
196 In the laboratory, plant samples were washed with ultra-pure water (Merck-Millipore Milli-
197 Q), placed into paper bags and dried in an oven at 60°C for one week. Soil samples (0.5-1.0
198 kg each) were dried at 60°C until constant weight. Dried samples (plant tissues and soil)
199 were weighed and mechanically grounded using a stainless steel grinder (particle diameter

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4 200 100 µm) for digestion. In the case of *A. compressus*, experiments were carried out only in
5 201 roots and leaves, since it lacks of a real stem. *A. compressus* shows a pseudo stem formed
6 202 by tightly packed overlapping leaf sheaths, like in other members of *Poaceae* family.
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10 204 2.3. Soil physicochemical parameters

11 205 Soil pH was determined on a soil:water mixture (1:2.5) by a potentiometric method
12 206 (Jaworska et al., 2016). Soil organic matter (SOM) was determined by the Walkley &
13 207 Black method (Rial et al., 2016; Walkley and Black, 1934). Available phosphorus (P),
14 208 inorganic nitrogen (NH₄) and potassium (K), were extracted in an Olsen modified extract
15 209 pH 8.5 (Olsen and Sommers, 1982), and quantified by photolorimetry-blue
16 210 phosphomolybdate (P), photolorimetry-blue indofenol (NH₄), and atomic absorption
17 211 spectroscopy (AAS). Cationic exchange capacity (CEC) was measured in ammonium
18 212 acetate buffered at pH 7 with barium chloride, and finally soil texture was determined by
19 213 the method of Bouyucos (Day, 1982). All measurements were performed on seven
20 214 replicates.
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26 216 2.4. Hg quantification in soil and plant samples

27 217 Dried samples were weighed, 0.2 g of each plant tissue (roots, stem and leaves) per
28 218 individual and 1 g of rhizosphere soil, and were left to soak in aqua regia, i.e. a mixture of
29 219 HCl and HNO₃ in a 3:1 ratio (v/v), for a week. Next, they were digested on an open heat
30 220 block (environmental express 54 Hot block SC154) for 2 h. After cooling, the samples were
31 221 diluted to 100 ml with HCl 0.1 M and stored until metal determination. Total Hg
32 222 concentration was determined following the hydride-generation technique in an atomic
33 223 absorption spectrophotometer (Perkin-Elmer, AANALYST-400). A Hg standard calibration
34 224 curve (100, 200, and 300 µg l⁻¹) was prepared in 10 ml of an acid mixture containing 1.5%
35 225 HNO₃ by triplicates. Two blank samples were also run simultaneously to estimate
36 226 background metal contamination from the digestion procedure. For each sample, 10 ml of
37 227 acid mixture containing 1.5% HNO₃ were added to 5 mL of the digestion mixture (prepared
38 228 by triplicates). Hg was determined using an aqueous solution of 3% (w/v) NaBH₄ in a 1%
39 229 (w/v) NaOH solution freshly prepared and filtered as reducing agent. An electrodeless
40 230 discharge lamp was used (Olmedo et al., 2013). Analytical grade chemical reagents and
41 231 highly purified deionized water were always used.
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50 233 2.5. Hg phytoextraction capacity calculation

51 234 To evaluate the Hg phytoextraction capacity of selected plant species, the bioaccumulation
52 235 factor (BF) was calculated as the ratio between the Hg concentrations in 3 plant tissues
53 236 (root, stem, leaves) and those in its corresponding rhizosphere soil sample: $BF = C_{\text{tissue}} / C_{\text{soil}}$
54 237 (González and González-Chávez, 2006). C_{tissue} and C_{soil} are Hg concentrations (mg Hg kg⁻¹
55 238 of dry weight). To evaluate the plants ability to transfer Hg from soil to their aerial parts,
56 239 the translocation factor (TF) was calculated as the ratio between Hg concentrations in aerial
57 240 plant parts (leaves and stem) and those in the plant root: $TF = C_{\text{aerial}} / C_{\text{roots}}$ (Chopin et al.,
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241 2008; Conesa et al., 2006). C_{aerial} and C_{roots} are Hg concentrations (mg Hg kg⁻¹ of dry
242 weight).

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244 *2.6. Quantification of AMF colonization and spores morphotypes*

245 Root samples from the three species were collected at the NZ and the IZ to determine the
246 presence of AMF colonization (%AMF_{col}). Roots were washed thoroughly with running
247 water to remove organic debris and soil particles and cut into small segments of 1 cm in
248 length. AMF colonization was assessed in each root segment following the method of
249 (Giovannetti and Mosse, 1980). Briefly, root segments were stained with 0.05% trypan
250 blue in lactophenol, and then washed with clear lactophenol to remove the excess of the
251 colorant. A total of 25 stained root segments were randomly selected for each plant species
252 and mounted on microscopic slides (five segments per slide). AMF colonization was then
253 assessed microscopically based on the presence of a blue stain. The percentage of root
254 colonization was calculated per plant species using the ratio of effectively stained segments
255 (presence of colonization) to the total of microscopically analyzed root segments for the
256 particular species per 100. Measurements were performed for three replicates of each plant
257 species in two sampling zones.

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259 AMF morphotypes associated with the roots of *E. polymnioides* and *M. zamorensis* were
260 isolated through the establishment of trap cultures. Five rizospheric soil samples per plant
261 species were placed in 2 l pots planted with a highly colonizable plant host (*Avena sativa*
262 L.) during five months. To fill each pot, fresh rizospheric soil samples were 1:1 mixed with
263 sterilized quartzitic sand (steam-sterilized sand at 100°C for periods of two hours each).
264 After planting, *A. sativa* plants were subjected to a regime of irrigation suspension for two
265 weeks to further promote AMF sporulation (Morton et al., 1995). AMF spores that
266 occurred in trap pots after five months were isolated by sieving 50 g of soil followed by
267 centrifugation in sucrose, as recommended by Sieverding et al. (1991). Pools of spores
268 sharing the same morphology were created under a stereomicroscope (40x) and further
269 separated into morphotypes by color and size. Photographs of each morphotype were
270 obtained from permanent microscope slides prepared using polyvinyl alcohol-lactic acid-
271 glycerin (PVLG) (Koske and Tessier, 1983) as mounting media under a compound
272 microscope. Morphotypes were morphologically described based on the number, size and
273 ornamentation of spore walls, reaction to Melzer reagent (iodine-potassium iodide-chloral
274 hydrate), and other criteria such as type of hyphal attachments and cicatrix if present.
275 Taxonomic identification down to genera level and to species level when possible was
276 performed according to an on-line AMF species catalogue and descriptions available at the
277 International Culture Collection of Mycorrhizal Fungi web site
278 (<http://invam.caf.wvu.edu/index.html>) following the classification given by some authors
279 (Redecker et al., 2013).

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4 281 *2.7. Statistical analysis*

5 282 SPSS version 20 software was used to carry out t-tests and ANOVA tests. The t test was
6 283 used to determine whether there were significant differences between two means
7 284 corresponding to two groups. For example, t test was used to elucidate if there were
8 285 significant differences between both zones in soil physicochemical parameters. The
9 286 ANOVA test was used to establish significant differences between more than two means,
10 287 corresponding to several groups. It allows drawing conclusions about Hg concentrations in
11 288 plant tissues, bioaccumulation and translocation factors and AMF colonization and
12 289 richness.
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18 291 **3. Results**

19 292 *3.1. Soil physicochemical parameters and Hg concentrations*

20 293 Acidic pH, high Fe and low N, P and K concentrations were recorded in all soil samples
21 294 from both the NZ and the IZ (Table 1). There were small differences between soils from the
22 295 sampling zones on most of the assayed parameters with the exception of N, and as
23 296 expected, Hg concentrations ($p < 0.05$). N and Hg were found at higher concentrations in
24 297 soil samples from the IZ.
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29 299 **Table 1.** Soil parameters from the sampling zones. NZ: natural mining zone; IZ: intensive
30 300 mining zone. Values are expressed as mean of seven independent measurements \pm standard
31 301 deviation

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Soil parameter	Sampling zone	
	NZ	IZ
pH (H ₂ O)	4.3 \pm 0.5	4.1 \pm 0.5
SOM (%)	2.8 \pm 0.9	2.4 \pm 0.5
N (mg kg ⁻¹)	44.8 \pm 6.5	85.3 \pm 10.5
P (mg kg ⁻¹)	6.9 \pm 0.9	8.6 \pm 2.3
K (cmol kg ⁻¹)	0.14 \pm 0.03	0.12 \pm 0.07
CEC (cmol kg ⁻¹)	6.3 \pm 1.4	6.9 \pm 1.2
Fe (mg kg ⁻¹)	505 \pm 154	664 \pm 207
Hg (mg kg ⁻¹)	0.19 \pm 0.09	4.8 \pm 1.2
S (mg kg ⁻¹)	33.2 \pm 3.9	123.1 \pm 41.5

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55 303 As it can be seen in Table 1, sulfur is an important component of the acidic soils we
56 304 studied. Sulfur concentration in the IZ is almost four times that of the NZ. The substantial
57 305 increase of sulfur in the IZ compared to the NZ reflects the gold mining practices in the IZ.
58 306 In addition, it is important to discuss the potential implications of high sulfur contents for
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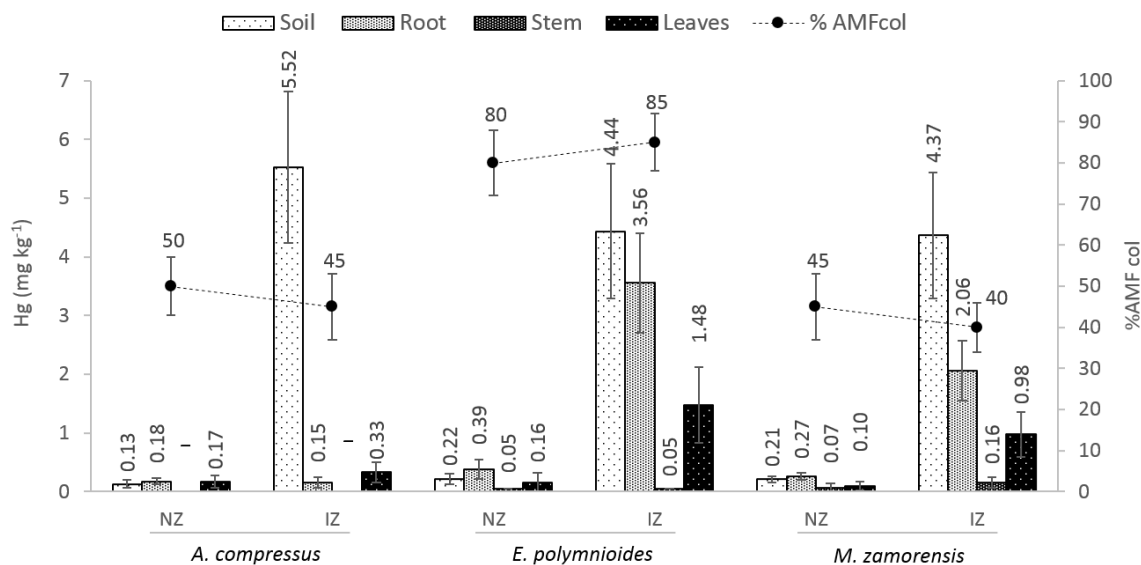
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307 Hg phytoremediation. Moreno et al. (2004) reported that sulfur in soil stimulates the
 308 accumulation of Hg in plant tissues. For example, *Brassica juncea* can concentrate Hg to 40
 309 mg kg⁻¹ in the plant canopy tissue after application of ammonium thiosulfate on mine waste
 310 contaminated with 2.8 mg Hg kg⁻¹. In fact, sulfur-mediated stimulation of Hg
 311 bioaccumulation has been reported by several researchers (Boudou et al., 1991; Moreno et
 312 al., 2005; Muddarisna et al., 2013; Wang et al., 2014).

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314 **3.2. Hg concentrations in plants tissues**

315 Hg concentrations in tissues (root, stem and leaves, for *E. polymnioides* and *M. zamorensis*
 316 and stem and root for *A. compressus*) from the studied species are shown in Figure 2. The
 317 highest values of Hg in roots and leaves of *E. polymnioides* and *M. zamorensis* were
 318 recorded in the IZ compared to the NZ as expected (p<0.05). Only in the stems the
 319 differences were not significant (p>0.05). Surprisingly, concentrations of Hg in the tissues
 320 of *A. compressus* were not statistically significantly different in both zones (p>0.05).

321
322 The highest averages for Hg accumulation in roots and leaves were recorded in *E.*
 323 *polymnioides* individuals from the IZ (3.56 mg kg⁻¹ and 1.48 mg kg⁻¹ respectively), followed
 324 by *M. zamorensis* individuals from the IZ (2.06 mg kg⁻¹ and 0.98 mg kg⁻¹ respectively).
 325 Considering the IZ, the leaves of *E. polymnioides* showed a significant difference compared
 326 to *A. compressus* (p<0.05) but insignificant to *M. zamorensis* (p>0.05). In the case of roots,
 327 concentrations of Hg in *E. polymnioides* were significantly different than concentrations in
 328 roots of both plants (p<0.05). This fact can ascribes a superior potential to *E. polymnioides*
 329 than *M. zamorensis* and *A. compressus* to accumulate Hg in soils with high Hg
 330 concentrations.



332
333 **Figure 2.** Hg concentrations (mg Hg kg⁻¹ of dry weight) in plants tissues (root, stem and
 334 leaves), and percentage of root arbuscular mycorrhizal fungal colonization (%AMF_{col}) on

individuals of each of the studied species collected at the natural mining zone (NZ) and the intensive mining zone (IZ). Values are expressed as mean of seven independent measurements \pm standard deviation.

3.3. Bioaccumulation and translocation factors

Bioaccumulation factors were generally higher in the NZ thanks to the lower Hg concentrations in the soil. Focusing on the IZ, roots seemed to be the preferred tissue for Hg bioaccumulation, since they concentrated the top bioaccumulation factors of all tissues on two of the three species studied. The highest bioaccumulation factor corresponded to the roots of *E. polymnioides* (0.80) even though the concentration of Hg in the soil was as high as 4.4 mg kg^{-1} .

A part of Hg looked to be further pumped up to the leaves in *E. polymnioides* and *M. zamorensis* (Figure 3) according to leaves' translocation factors over 0.3. Nevertheless, *A. compressus* translocated a higher proportion of Hg from the roots to the leaves and it accounted for the largest leaves/root translocation factor.

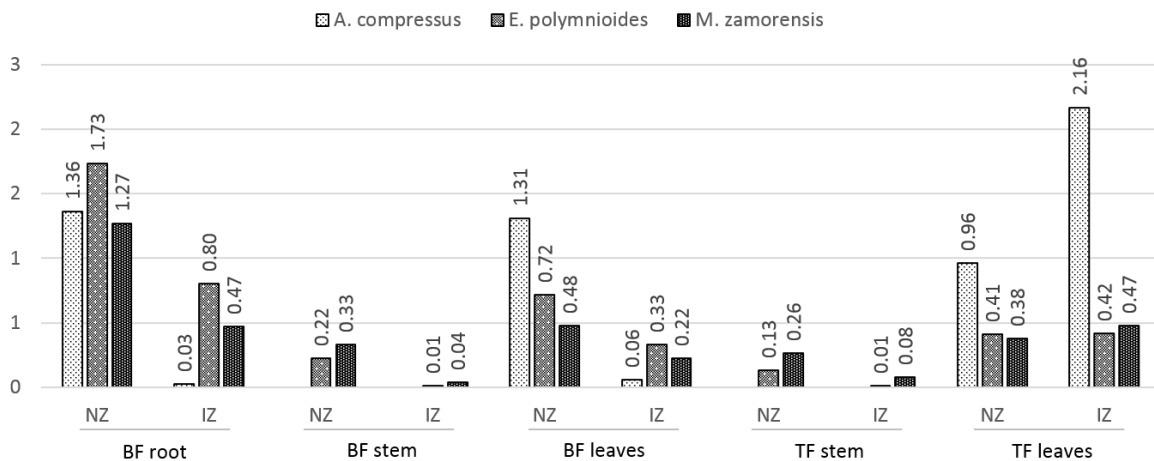


Figure 3. Bioaccumulation factor (ratio of the plant tissue to soil metal concentration, BF) and translocation factor (ratio of the stem and leaves to root metal concentration, TF) for Hg in three native plants from natural mining zone (NZ) and intensive mining zone (IZ). Values are calculated based on the corresponding mean values in Figure 2.

3.4. AMF colonization and richness analyses

All plants showed AMF colonization in their roots (see %AMF_{col} in Figure 2). *E. polymnioides* showed by far the highest root colonization by AMF ($p < 0.05$). Interestingly, root colonization in *E. polymnioides* was higher in the IZ than in the NZ. The opposite was observed in the other two species. Since *A. compressus* showed the lowest values of Hg accumulation (Figure 2) and mycorrhizal colonization, the analysis of AMF richness was circumscribed to *E. polymnioides* and *M. zamorensis*. Seven morphotypes from the

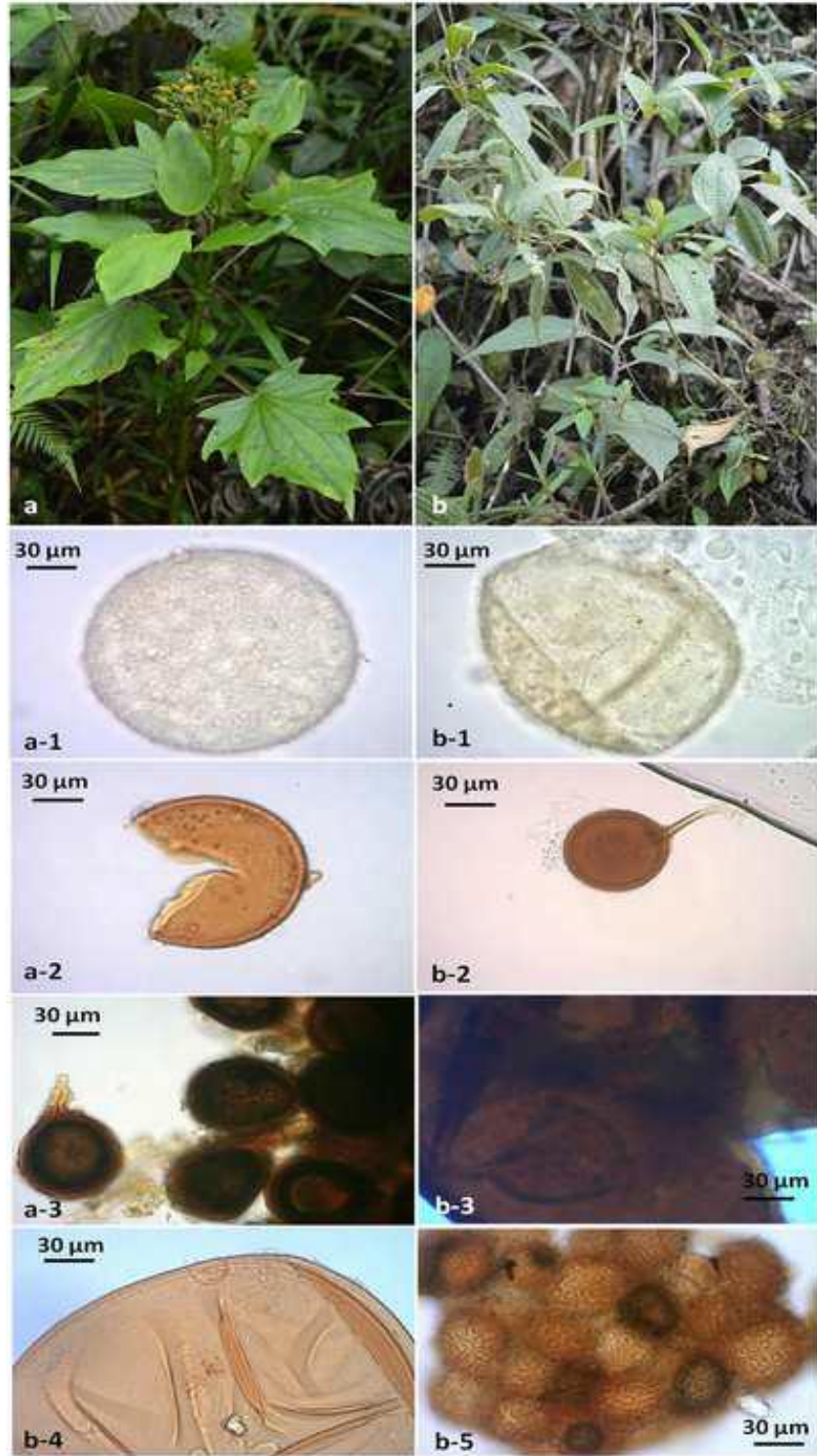
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365 following genera were identified in AMF morphotype based characterization: *Glomus*,
366 *Acaulospora*, *Ambispora* and *Racocetra* (Figure 4). Three of these morphotypes
367 (*Ambispora* sp., *Glomus* sp. 1 and *Glomus* sp. 2) were found associated to *E. polymnioides*
368 while five morphotypes (*Ambispora* sp., *Glomus* sp. 3, *Racocetra* cf. *gregaria*,
369 *Acaulospora* cf. *capsicula* and *Sclerocystis sinuosa*) were associated to *M. zamorensis*.
370 Only one morphotype (*Ambispora* sp.) was found in common between both plant species.

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373 **Figure 4. Arbuscular mycorrhizal fungi morphotypes associated with the rhizospheric**
374 **soil of a) *Erato polymnioides* (morphotypes a-1, a-2, a-3) and b) *Miconia zamorensis***
375 **(morphotypes b-1, b-2, b-3, b-4, b-5). a-1) *Ambispora* sp. a-2) *Glomus* sp. 1, a-3) *Glomus***
376 **sp. 2, b-1) *Ambispora* sp. b-2) *Glomus* sp. 3, b-3) *Racocetra* cf. *gregaria* (sporogenous cell)**
377 **b-4) *Acaulospora* cf. *capsicula* (cicatrix), b-5) *Sclerocystis sinuosa*.**
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4. Discussion

Chemical properties observed in Chinapintza soil samples were those typical of tropical humid forest soils: acidic pH and low P, N and K concentrations. These results are in agreement with other analyses conducted in the region (Jiménez et al., 2007). Hg concentrations recorded at both sampling zones (NZ: 0.19 mg kg⁻¹; IZ: 4.8 mg kg⁻¹) exceed the level reported for regular crustal abundance (0.01-0.05 mg kg⁻¹) (Rudnick and Gao, 2003), and the threshold for polluted soils in Ecuador (0.1 mg kg⁻¹) (Ecuadorian Ministry of the Environment, 2015). Hg concentration in the IZ is similar to the upper-end value reported at ASGM areas in Bolivia (0.5-2.4 mg kg⁻¹) (Terán-Mita et al., 2013). However, Hg in the IZ is far lower than polluted mining sites in Asia, where Hg reaches 44 mg kg⁻¹ (Li et al., 2013) and 35 mg kg⁻¹ (Zhang et al., 2010) in China.

Soil pHs, electric conductivity and sulphur in polluted soils showed influence on plant growth and correlation with phytoremediation in previous studies (Adamczyk-Szabela et al., 2015; Salimi et al., 2012). High electric conductivity levels of irrigation water can affect to the absorption of metals by the plant (Salimi et al., 2012). On the other hand, when the pH is lower the metal ions show greater cation exchange capacity and become more available in the aqueous medium, making the metal more bioavailable to plants, as demonstrated in *Valeriana officinalis* (Adamczyk-Szabela et al., 2015). The most intensive germination in *Valeriana officinalis* and plant growth was observed at pH=5.1. Increased alkalinity to pH=10 significantly inhibited both the seed germination and plant growth. Valerian seeds inoculated in soil characterized by a low or very high pH values (3.5 and 13.0) failed to germinate even in non-contaminated soils, showing the negative influence of too low or too high pH value.

Soil Hg concentration in the NZ remains higher than the Ecuadorian nation-wide reference value of 0.1 mg kg⁻¹, highlighting the slow mobility of Hg in terrestrial ecosystems (Hintelmann et al., 2002) and the need of effective extraction methods. Slow mobility of Hg in polluted soils is thought to be a consequence of the presence of no readably bioavailable chemical species, e.g. Hg-sulfide and elemental Hg (Piani et al., 2013). Therefore, it is not accumulated in plant tissues in significant amounts. Hg accumulation relies on plants ability to absorb and transport the Hg-thiosulfate complex (Wang et al., 2012). The ability to convert Hg-sulfide into Hg-thiosulfate has been reported for a number of sulfur-oxidizing bacteria (Vázquez-Rodríguez et al., 2015), which are responsible for making Hg bioavailable for plants. However, Hg-sulfide oxidation and plant Hg-thiosulfate uptake are not necessarily coupled, since the first process might occur far from plant roots. AMF associated with plant roots may improve plant uptake of bioavailable forms of Hg (Ekamawanti et al., 2014; Pichardo et al., 2012; Selvaraj et al., 2005). In soils like those in Chinapintza, in which phosphorous concentration is relatively low, AMF colonization is expected to be favoured since it brings several benefits for plants to fulfilling nutrient requirements (Ryan et al., 2000). Altogether, these observations could explain why the

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421 results of this study suggest a relationship between the amount of AMF colonization and
422 the species' ability to accumulate Hg in their tissues. *E. polymnioides* showed the highest
423 Hg accumulation capacity as well as the highest percentage of AMF colonization. High
424 percentages of AMF colonization in heavy metal hyperaccumulating plants of the
425 Asteraceae family (the one to which *E. polymnioides* belong) have been reported (Turnau
426 and Mesjasz-Przybylowicz, 2003).

427
428 In fact, *E. polymnioides* is the only species in this study that showed no saturation in its Hg
429 accumulation capacity when growing in the IZ. Furthermore, it was the only species
430 displaying an increase in AMF colonization in the IZ compared to the NZ. Both *A.*
431 *compressus* and *M. zamorensis* showed clear signs of Hg accumulation capacity saturation
432 when growing at concentrations of Hg as high as 4.8 mg kg⁻¹ in the IZ and a reduction of
433 AMF colonization compared to the NZ. Hg accumulation capacity saturation has been
434 reported in other plant species, such as *Jatropha curcas* (Euphorbiaceae family), at Hg
435 concentrations in the range of those in the IZ (Marrugo-Negrete et al., 2015). Asteraceae
436 family shows advantages, such as Hg tolerance and efficiency to accumulate Hg in the
437 polluted soil as it behaves as a hyperaccumulator. Also, it was identified as the most
438 common species in the study sites showing Hg as the major metal distribution in the soil.

439
440 Interestingly, only three AMF morphotypes were found associated to *E. polymnioides*. This
441 might suggest that Hg accumulation is independent of AMF diversity. Instead, Hg
442 accumulation could be based on a core of highly infective strains of AMF capable of
443 enhancing plant nutritional status and making Hg-bioavailable species accessible to plant
444 tissues as a side effect. A recent study has demonstrated that native AMF adapted to soil
445 polluted conditions can effectively increase the total content of metals present in plant
446 tissues as well as plant root biomass production (Ouanyi et al., 2016). On the other hand,
447 the involvement of AMF in phytoremediation may also occur through specific mechanisms
448 of metal retention and immobilization in the mycelium of the fungus, reducing its
449 translocation to the stem (Cabral et al., 2015; Miransari, 2011). The production of
450 metallothioneins by AMF increases the deposition of heavy metals in their cells (Miransari,
451 2011). In this sense, this symbiosis protects the plant due to the stabilization of metals in
452 the radical compartment, which is known as phytostabilization mediated by
453 microorganisms (Meier et al., 2012). AMF-mediated phytostabilization has also been
454 proposed as an explanation for the decrease in Hg phytovolatilization in mycorrhizal plants
455 compared to non-mycorrhizal plants (Yu et al., 2010).

456
457 Values of Hg found in roots of *E. polymnioides* individuals from the IZ (3.56 mg kg⁻¹)
458 represent as much as the 74% relative to rhizosphere soil Hg concentration in the IZ. These
459 values are comparable to those reported for other members of the Asteraceae family:
460 *Matricaria recutita* (1.57 mg kg⁻¹ in aerial tissue) and *Scolymus hispanicus* (2.72 mg kg⁻¹),
461 growing in polluted soils with 5.53 mg kg⁻¹ of Hg (Millán et al., 2006). Considering that

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462 0.5-1.0 mg Hg kg⁻¹ in plant tissues can produce growth depression (Kabata-Pendias, 2011),
463 *E. polymnioides* showed an elevated tolerance to this heavy metal. In addition, *E.*
464 *polymnioides* displayed the least root-to-leaves Hg translocation, in contrast to *A.*
465 *compressus*, which has the ability to translocate an important fraction of the accumulated
466 Hg from the roots to its leaves, especially at the IZ (2.16). Based on these findings, *E.*
467 *polymnioides* could be seen mainly as a Hg stabilizer species while *A. compressus* capable
468 of phytoextracting Hg (incorporation and redistribution to the aerial parts of the plant).

469
470 Regardless the final goal of a remediation strategy, heavy metal phytoextraction is desirable
471 in stages where appropriately cutting and final disposal of the aerial plant biomass has been
472 previously covered. In contrast, phytostabilization could be more favorable in less
473 accessible locations, where plant roots and rhizospheric microbial interactions can
474 immobilize Hg, reducing its incorporation into ground and/or water bodies and food chain
475 (Sarma, 2011). In a geographical setting like that of Chinapintza, with limited accessibility,
476 highly dense tropical forests and many ASGM sites around a water stream that goes
477 directly into downstream communities, Hg phytostabilization is preferred over
478 phytoextraction. A highly efficient accumulator and tolerant plant such as *E. polymnioides*
479 could represent an attractive strategy for the environmental remediation of Chinapintza
480 soils affected by ASGM. Even more, considering it is native to the area, an increase in its
481 population size for remediation purposes should not represent an ecological treat

482
483 On the other hand, volatilization of Hg⁰ by transpiration is a possible route for Hg
484 detoxification in plants (Moreno et al., 2008). Future studies are essential to evaluate the
485 phytovolatilization of Hg and the possible contribution to air pollution of this native plant.

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488 **5. Conclusions**

489 *E. polymnioides* (Asteraceae family) showed high potential as an Hg hyperaccumulator
490 plant with higher affinity of mycorrhizal association helpful to survive against metal
491 toxicity. Thus, it is a suitable candidate to be considered as an Hg hyperaccumulator when
492 designing environmental friendly and sustainable strategies for Hg remediation at ASGM
493 areas. A deeper understanding of AMF role in Hg phytoextraction should be paid a special
494 attention in future studies. The availability of *E. polymnioides*, together with the associated
495 AMF diversity isolated in the present study, could be instrumental for this task.

496
497 **Acknowledgements**

498 This work had the financial support of: (1) National Secretary of Higher Education,
499 Science, Technology and Innovation of the Republic of Ecuador (SENESCYT from its
500 acronym in Spanish) in the frame of the Prometeo Project; (2) UTPL SmartLand initiative,
501 research program PROY_CCNN_1138.

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Figure

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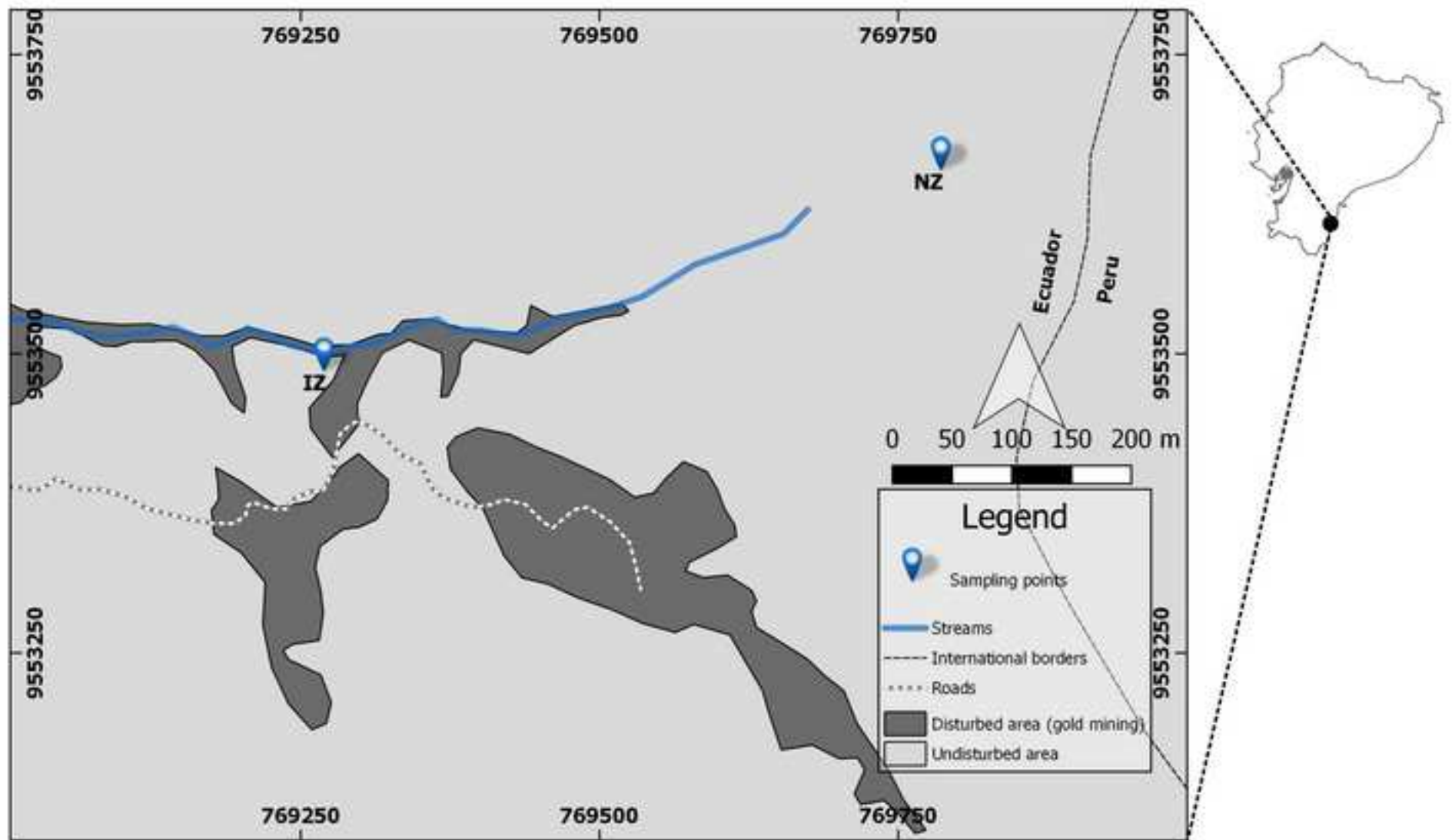


Figure2

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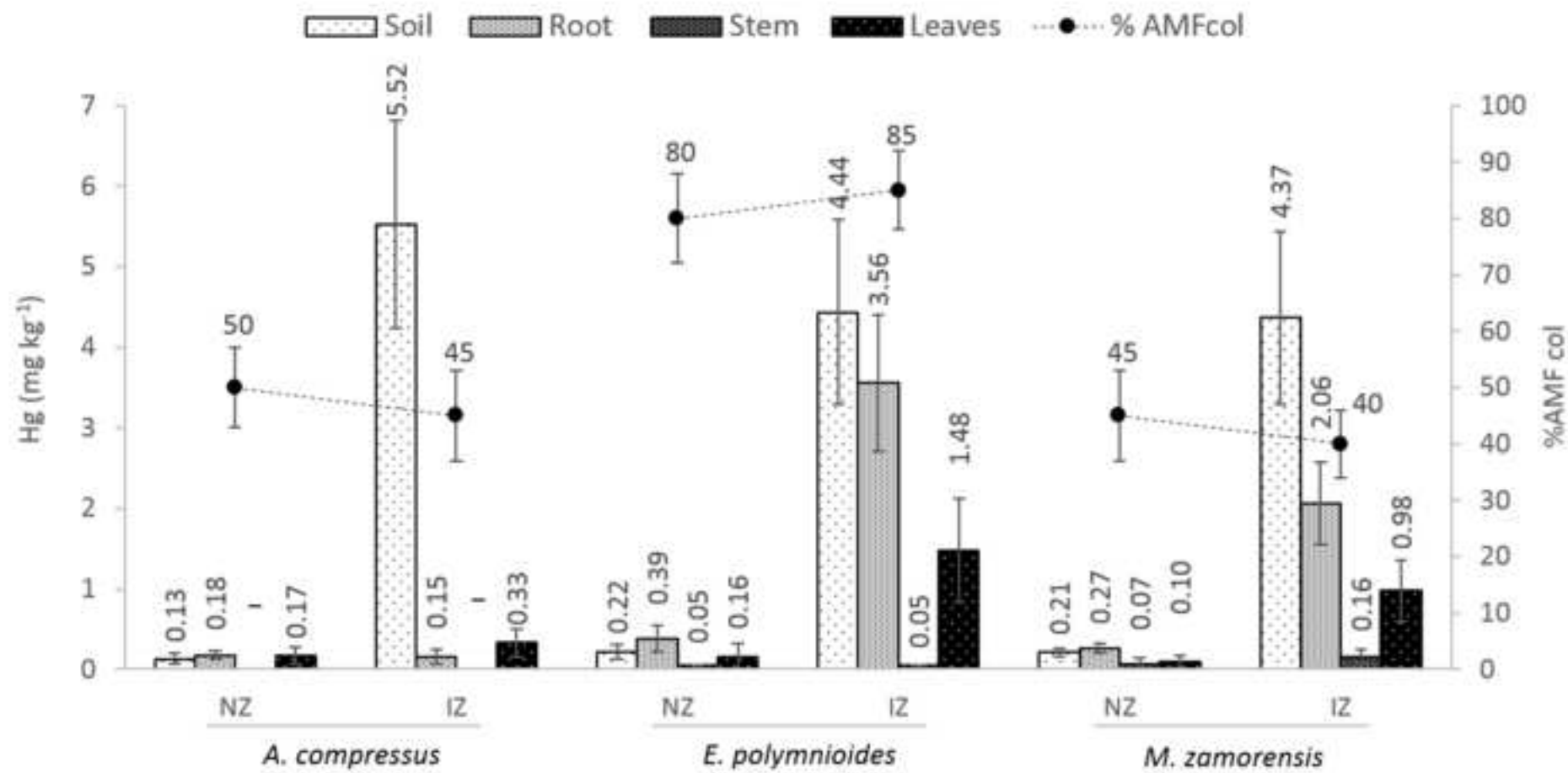


Figure3

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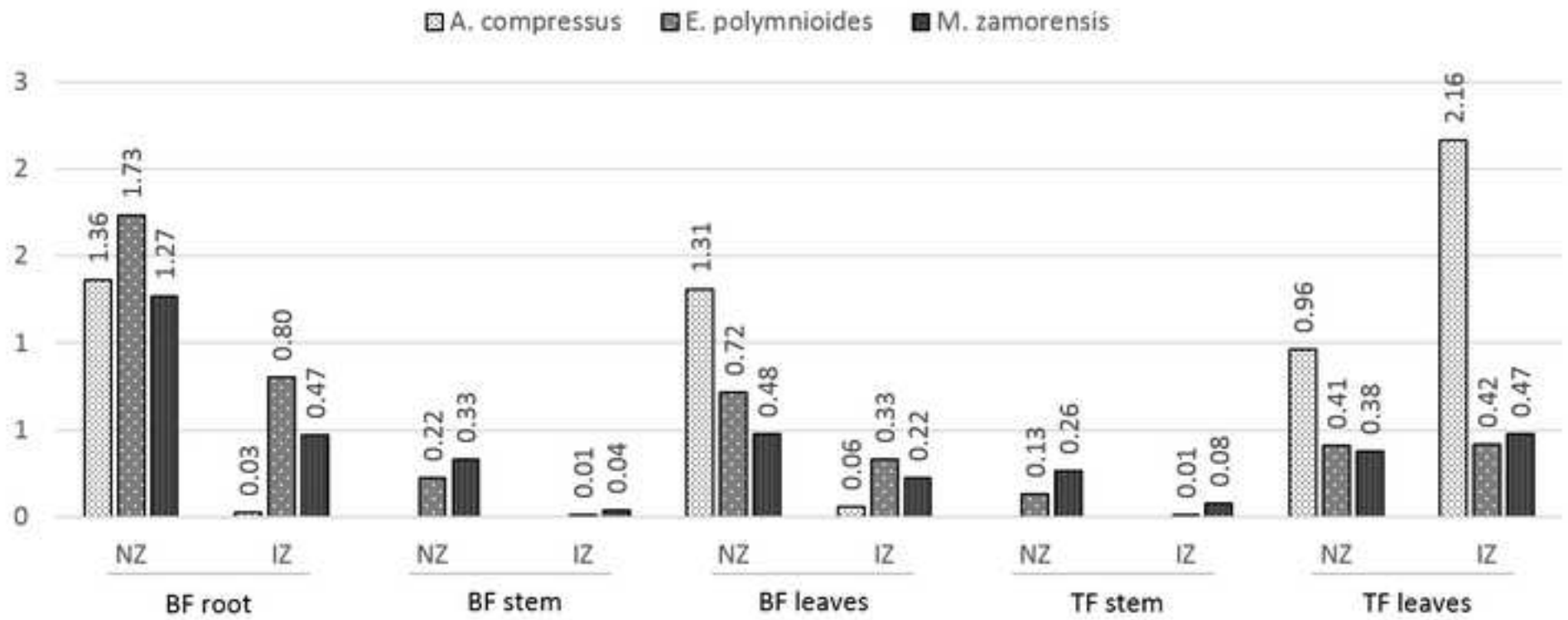


Figure4

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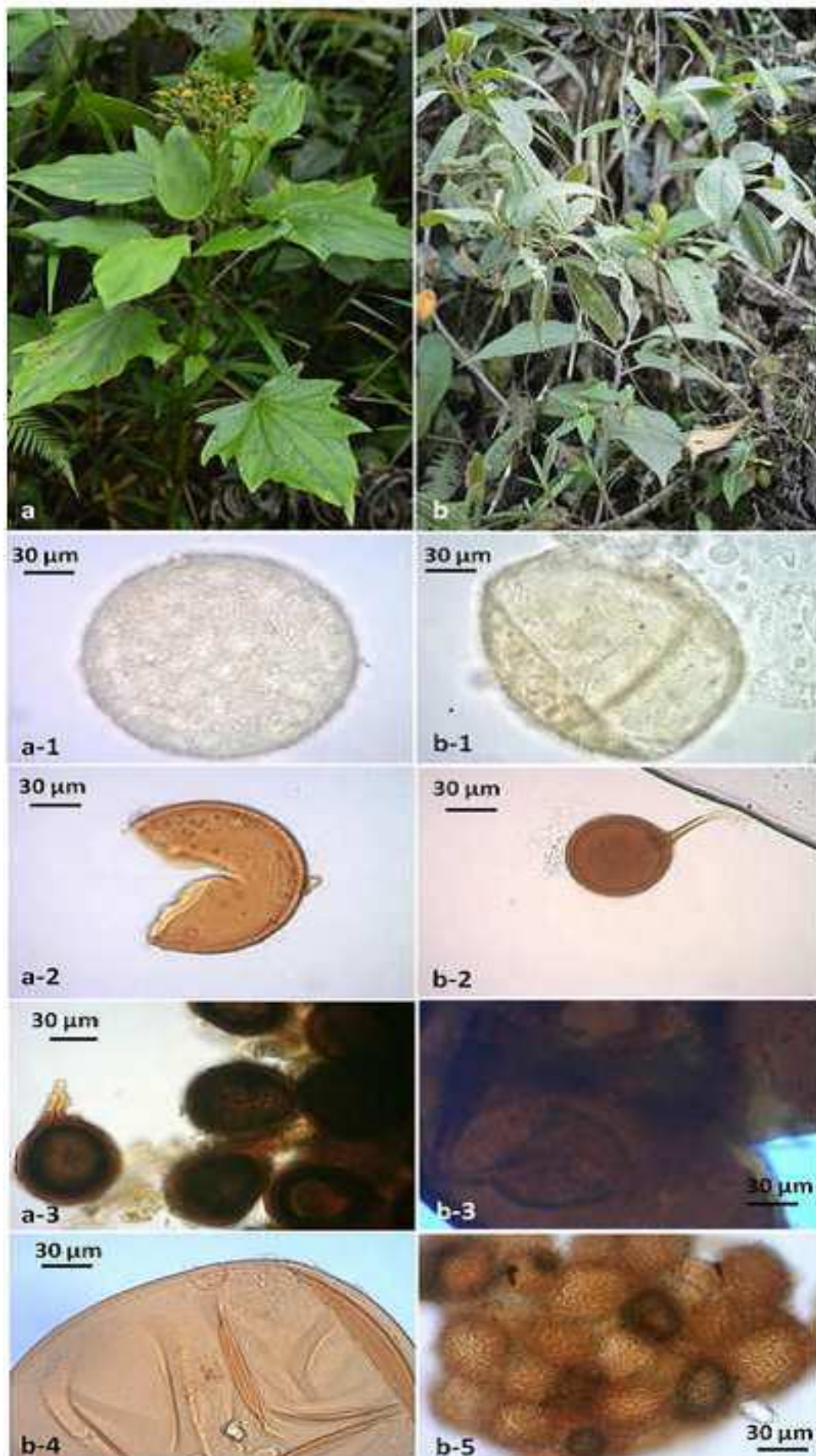


Table 1. Soil parameters from the sampling zones. NZ: natural mining zone; IZ: intensive mining zone. Values are expressed as mean of seven independent measurements \pm standard deviation

Soil parameter	Sampling zone	
	NZ	IZ
pH (H ₂ O)	4.3 \pm 0.5	4.1 \pm 0.5
SOM (%)	2.8 \pm 0.9	2.4 \pm 0.5
N (mg kg ⁻¹)	44.8 \pm 6.5	85.3 \pm 10.5
P (mg kg ⁻¹)	6.9 \pm 0.9	8.6 \pm 2.3
K (cmol kg ⁻¹)	0.14 \pm 0.03	0.12 \pm 0.07
CEC (cmol kg ⁻¹)	6.3 \pm 1.4	6.9 \pm 1.2
Fe (mg kg ⁻¹)	505 \pm 154	664 \pm 207
Hg (mg kg ⁻¹)	0.19 \pm 0.09	4.8 \pm 1.2
S (mg kg ⁻¹)	33.2 \pm 3.9	123.1 \pm 41.5